

Investigation of heavy metals emission from pyrolysis product of rubber wastes treated with ashing

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Abstract

© 2018 BRNSS Publication Hub. All rights reserved. Aim: In this paper, the emission of heavy metal ions from potential sorption materials is investigated. Samples were obtained by low-temperature pyrolysis from rubber waste (used car tires) and treated with dry ashing. The use of a solid pyrolysis product after ashing involves the purification of wastewater from contaminants. Methods: The content of heavy metal ions in aqueous extraction and extraction with an acetate-ammonium buffer of a solid product of pyrolysis of rubber waste was determined by means of atomic emission spectrometry. Of decreasing mass concentration, the heavy metal ions in the aqueous extract of the solid pyrolysis product after treatment with ashing are arranged in the following order: Zn, Si, Mn, Sr, Co, Ba, Mo, Ni, and Sb. In the acetate-ammonium extract of the test sample, the heavy metals are arranged in the following order in the order of decreasing values: Zn, Mn, Co, Fe, Sr, Cu, Al, Ni, B, V, Pb, Cr, Ba, Se, Pb, and Sb. Results: The obtained results show that the aqueous extract of the solid pyrolysis product of rubber waste after treatment with "dry" ashing does not exceed the normative indices for sewage by the content of heavy metals. According to the values of the concentration coefficient relative to the permissible concentration of pollutants in the wastewater admitted to discharge into the centralized system of wastewater disposal, the excess of the norm takes place according to Zn. Conclusions: It has been established that the solid pyrolysis product of rubber waste treated with ashing does not pollute the wastewater. This implies the possibility of using the processed pyrolysis product from environmental positions with limiting the discharge of wash water directly into fishery water reservoirs.

Keywords

Adsorption, Ashing, Emission, Heavy metal ions, Pyrolysis, Rubber waste

References

- [1] Nasyrov IA, Zinnatov RR, Mavrin GV. Pyrolysis of silt Sediments of Sewage Treatment Plants as a Method of Utilization. Belgorod: Modern Trends in the Development of Science and technology: A collection of Scientific Papers on the Materials of the VIII International (correspondence) Scientific and Practical Conference; 2015. p. 19-20.
- [2] Gulyaev IS, Dyakov MS, Savinova YN, Glushankova IS. Analysis and study of methods of neutralization and disposal of sewage sludge biological treatment plant. Bull PNPU Prot Environ Transp Saf Life 2012;2:18-32.
- [3] Minori Uchimiya, Wartelle LH, Klasson KT, Fortier CA, Lima IM. Influence of pyrolysis temperature on biochar property and function as a heavy metal sorbent in soil. J Agric Food Chem 2011;59:2501-10.

- [4] Nasyrov IA, Ahmetov VM, Miftahov MN, Mavrin GV, Sokolov MP. The problem of disposal sludge treatment plants. *Int J Pharm Technol* 2016;8:14359-65.
- [5] Deliyanni EA, Peleka EN, Matis KA. Modeling the sorption of metal ions from aqueous solution by iron-based adsorbents. *J Hazard Mater* 2009;172:550-8.
- [6] Zaini MA, Okayama R, Machida M. Adsorption of aqueous metal ions on cattle-manure-compost based activated carbons. *J Hazard Mater* 2009;170:1119-24.
- [7] Nasyrov IA, Dvoryak SV, Shaikhiev IG. Sorption properties of carbon waste pyrolysis product for biological wastewater treatment. *Acta Technica* 2016;61:307-14.
- [8] Nasyrov IA, Ahmadiyeva AI, Fazullin DD, Mavrin GV, Sokolov MP. Petroleum containing wastewater products purification by carbon-containing wastes pyrolysis products. *Turk Online J Des Art Commun* 2017;7:1713-28.
- [9] Nasyrov IA, Mavrin GV, Ahmetshina AR, Ahmadiyeva AI. Sorption properties of pyrolysis products of sludge, wood waste and rubber waste for heavy metal ions. *J Fundam Appl Sci* 2017;9 S1:1615-25.
- [10] Bepamyatnov GP. Threshold limit values of harmful substances in air and water. The handbook for the choice and hygienic assessment of methods of neutralization of industrial wastes. 2nd PROD. The Lane and Additional-L.: Chemistry. 2016; 3:673-692.